



Course Code:	1022SCG
Course Name:	Chemistry 1B
Trimester:	Trimester 1, 2019
Program:	Diploma of Science
Credit Points:	10
Course Coordinator:	Dr Gretel Heber
Document modified:	20 <sup>th</sup> February 2019

### Teaching Team

Your lecturer/tutor can be contacted via the email system on the portal.

Name	Email
Dr Gretel Heber	Gretel.Heber@staff.griffithcollege.edu.au
Dr Carolyn Munce	Carolyn.Munce@staff.griffithcollege.edu.au

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### Staff Consultation

Your lecturer/tutor is available each week for consultation outside of normal class times. Times that your lecturer/tutor will be available for consultation will be given in the first week of lectures. A list of times and rooms will be published on the Griffith College Portal under the “myTimetable” link.

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## Prerequisites

To successfully enrol in this Course, you must provide evidence that you have completed one of the following courses:

- 1021SCG - Chemistry 1A
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## Brief Course Description

This course covers a range of basic physical and chemical concepts relevant to the study of the biosciences. Topics include basic organic chemistry with emphasis on functional group chemistry and its relevance to the functional properties; physical chemistry and spectroscopy; basic inorganic chemistry, focusing on the properties of p-block elements and transition metal complexes.

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## Rationale

Chemistry is known as the central science in that it is important to our understanding interactions that occur in all the other scientific disciplines. A knowledge and understanding of chemistry and how it interacts with biological processes is essential in understanding the chemical, physical and structural bases upon which living organisms are constructed and the biological properties that emerge in living organisms. At the foundation of modern biology including human biology and medicine, is the study of cells, the smallest unit of living organisms and the study of molecular biology, the chemistry of living organisms. With this knowledge comes the ability to make predictions about living processes and the manipulation of biological processes, such as in the treatment of human diseases or the development of applications in Biotechnology. Consequently, Chemistry 1B is a core course in the study of the biological sciences.

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## Aims

The aim of the course is to motivate the fundamental concepts and methods of the chemical sciences that are essential to any further study in the modern science, biological, environmental, engineering, biotechnology and biomedical sciences. This course provides an introduction to the fundamental principles of organic, inorganic and physical chemistry relevant to further study in the sciences.

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## Learning Outcomes

Upon successful completion of this course students will be able to...

1. Identify and explain chemical concepts, reactions and properties in the organic, physical and inorganic chemistry fields,
2. demonstrate effective communication and sound critical

analysis skills in a range of contexts, along with well-developed laboratory skills and safe working practices in the laboratory.

3. demonstrate effective problem solving skills in the chemical sciences including: identifying functional groups, identifying stereoisomers and their properties, predicting the products of organic reactions, calculating the pH of weak acid/base and buffer solutions, interpreting NMR spectra, conducting redox calculations, identifying nuclear particles, constructing nuclear equations, solving radionuclide decay problems, identifying rate laws and mechanisms from experimental data, using crystal field theory to explain the structure and properties of transition metal complexes and linking chemical knowledge to concepts in other areas of science.

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Texts and Supporting Materials

### **Learning Resources**

You are provided with a variety of learning resources and teaching and learning activities to assist you in engaging this course and in gaining a thorough understanding of the course material. While all these resources are provided, you need to take responsibility for your own learning by first learning how to utilise these resources to maximum benefit in terms of developing your understanding of the course material. A brief guide to the resources and teaching is provided below.

#### **Required**

Blackman, Bottle, Schmid, Mocerino and Wille (2016), Chemistry 3<sup>rd</sup> ed., John Wiley & Sons, Australia as used for Chemistry 1A.

Lecture notes and supplementary material will be available on the portal.

The textbook should be used in conjunction with lecture notes and other materials provided in lectures and via the course web site as a major source of detailed information about the course material. It provides detailed diagrams, illustrations and problems that should be valuable aids in your learning.

#### **Supplementary notes:**

Supplementary notes, including lecture notes and tutorial questions for each module will be available to be downloaded from the portal during the trimester.

**Laboratory Manual:** Griffith College 1022SCG Chemistry 1B Laboratory Manual 2017 available from the Co-op Bookshop Mt Gravatt.

#### **Web Sites & Literature References:**

The course Web site is available via Griffith College Portal The Portal and email will be used by the convenor as the main means of communicating information to you about the running of the course e.g workshop quiz results, any timetable changes etc consult your email and the Griffith College Portal regularly.

Links to other sites are provided on the course site and also in the Study Guide. Your lecturer may provide further resources such as recommended literature references and other useful Web sites and these are to be used to further develop your understanding of particular topics.

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## Organisation and Teaching Strategies

The course consists of 4 Modules and a Laboratory Component.

### **MODULES TOPICS TEACHING & LEARNING ACTIVITIES**

Module 1 Weeks 1 - 4 Organic Chemistry 12 hr lectures, 4 tutorials, 4 workshops,

Module 2 Weeks 5 - 8 Physical chemistry and spectroscopy 12 hr lectures, 4 tutorials, 4 workshops,

Module 3 Weeks 9 - 11 Inorganic chemistry 9 hr lectures, 3 tutorials, 3 workshops.

Week 12 will be reserved for revision of the course content covered from week 1 to week 11.

**PLEASE NOTE:** Lab classes for this course will be 4 hours each over five weeks in room N44\_3.16A (Nathan Campus). Experiments in organic chemistry, inorganic chemistry, buffers, energetics and spectroscopy.

The Course is taught by using a variety of teaching and learning activities including lectures, and intensive, facilitated workshops, tutorials and laboratory activities. Course material will be presented in four (4) Learning Modules as detailed above.

The Teaching and Learning Activities used in the different modules are

#### **Lectures:**

For each module, lecture presentations will be provided highlighting the main points covered by the module topics and explaining the key concepts developed in the module. Detailed content is provided by the Textbook and other sources. The purpose of the lectures is to explain and discuss concepts based on the knowledge of content. To gain maximum benefit from the lectures you should have some general knowledge of that content before participating in the lecture presentation.

#### **Workshops:**

Workshops help to develop further understanding of course content through discussion of relevant topics and examples using a problem solving approach.

#### **IMPORTANT:**

To gain the most from this course, attendance at all workshops is strongly advised.

#### **Laboratory:**

For the Laboratory Component the class is divided into groups. You will do five laboratory sessions as per timetable. Laboratory attendance at **ALL** laboratory sessions is compulsory.

## *Class Contact Summary*

### **Attendance**

To gain maximum benefit from this course, it is important that you are present for all lectures and workshops.

### **Participation**

You are expected to have pre-read the lecture notes and relevant chapters in the textbook before coming to lectures each week and to actively participate in all class activities.

### **Consultation times**

Attendance during consultation times is optional but you are encouraged to use this extra help to improve your learning outcomes.

### **Laboratory Sessions**

Punctuality is always important but especially so for the laboratory classes. It is important that you arrive prepared for each laboratory session and are ready to start at the time given in the timetable. Failure to do so will result in a loss of marks and possible exclusion from that lab session.

Preparation for lab sessions involves reading through the relevant chapters in the laboratory manual and completing pre-lab exercises. You are also responsible for bringing your lab coat, safety glasses and wearing appropriate footwear that cover the whole foot. Marks will also be deducted for failure to come adequately prepared.

### **Independent Learning**

In addition to the 5 hours of contact each week, you are also expected to undertake a minimum of 6-7 hours each week in undertaking learning and project activities related to this course.

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## Content Schedule

### **Module 1: Organic Chemistry**

Bonding and structure identification in organic chemistry; basic organic chemistry with emphasis on functional group chemistry - alkanes, alkenes, alkynes, haloalkanes, aromatics, alcohols, carboxylic acids and derivatives, and amines. Stereochemistry and weak acid/base concepts are also covered.

### **Module 2: Physical chemistry and spectroscopy**

Revision of spontaneous processes, entropy, free energy. Introduction to electrochemistry and electrochemical cells. Introduction to reactions kinetics, catalysis and elementary reactions.

Introduction to spectroscopy for organic systems, especially  $^1\text{H}$  NMR,  $^{13}\text{C}$  NMR and mass spectroscopy. Nuclear chemistry, equations and carbon dating.

### Module 3: Inorganic Chemistry

Chemistry of the elements in groups 15 to 17 with emphasis on the period 2 and 3 elements (carbon and silicon; nitrogen and phosphorus, oxygen and sulphur). Introduction to transition metal coordination chemistry, and applications of a biological and environmental context.

### Laboratory Component

Five Laboratory sessions on Organic Chemistry, Inorganic Chemistry, Buffers, Physical chemistry and Spectroscopy

PLEASE NOTE: Lab classes for this course will be in room N44\_3.16A (Nathan Campus) as advised in your timetable.

### Weekly Teaching Schedule

Week	Topic	Activity	Readings
1	Module 1: organic chemistry	Lecture	Textbook ch. 16, 21; class notes
	Organic chemistry	Workshop	Textbook and study guide
2	Module 1: organic chemistry	Lecture	Textbook ch. 17, 18; class notes
	Organic chemistry	Workshop	
3	Module 1: organic chemistry	Lecture	Textbook ch. 11, 19; class notes
	Organic chemistry	Workshop	
	Module 1.1 online quiz	Online quiz 1	
4	Module 1: organic chemistry	Lecture	Textbook ch. 21, 23; class notes
	Organic chemistry	Workshop	
5	Module 2: Redox	Lecture	Textbook ch. 12 and class notes
	Organic chemistry	Workshop	
	Module 1.2 online quiz	Online quiz 2	
6	Module 2: Chemical kinetics	Lecture	Textbook ch. 15 and class notes
	Redox	Workshop	

	Module 1 in-class quiz	In-class quiz 1	
7	Module 2: Spectroscopy	Lecture	Textbook ch. 20 and class notes
	Kinetics	Workshop	
	Module 2.1 online quiz	Online quiz 3	
8	Module 2: Nuclear chemistry	Lecture	Textbook 27 and class notes
	Spectroscopy	Workshop	
9	Module 4: Inorganic chemistry	Lecture	Textbook ch 13 and class notes
	Nuclear chemistry	Workshop	
	Module 2.2 online quiz	Online quiz 4	
10	Module 4: Inorganic chemistry	Lecture	Textbook ch 13 and class notes
	Inorganic chemistry	Workshop	
	Module 2/3 in-class quiz	In-class quiz 2	
11	Module 4: Inorganic chemistry	Lecture	Textbook ch 14 and class notes
	Inorganic chemistry	Workshop	
12	Course revision	Lecture	

## Assessment

This section sets out the assessment requirements for this course.

### *Summary of Assessment*

Item	Assessment Task	Weighting	Relevant Learning Outcomes	Due Date
1	Online Quizzes 1 and 2, Module 1 Online Quizzes 3 and 4, Module 2	10%	1,3	3, 5, 7, 9
2	Laboratory <i>- Students must pass this assessment with a mark</i>	20%	1,2,3,4	Progressive 1-12

	<i>of at least 15 out of 25 to pass the course</i>			
3	In-class test 1, Module 1 (weeks 1-4) In-class test 2, Modules 2-3 (weeks 5-9)	12.5% 12.5%		6, 10
4	End of Trimester Exam  - <i>Students must pass this assessment with a mark of at least 40% to pass the course</i>	45%	1,2,3	14

### *Assessment Details*

#### **Online Quiz 1-4 (2.5% each)**

These quizzes will test students' understanding, interpretation and application of the relevant course materials and problem solving skills. The Wiley Plus system will be used for these online quizzes. Also, these quizzes will help students to prepare better before undertaking the in-class tests.

#### **In-class test 1 (12.5%) and test 2 (12.5%)**

These tests will assess students' understanding, interpretation and application of the relevant course materials and problem solving skills under normal exam conditions.

#### **Laboratory Assessment(20 %):**

Attendance and participation in **ALL** the Laboratory sessions is compulsory. Attendance and participation will be recorded.

Students must attend **ALL laboratory sessions** and gain an overall pass on the Laboratory component in order to gain any credit for the course.

For further instructions on the laboratory component and for details on laboratory assessment please refer to the separate "Griffith College 1022SCG Chemistry 1B Laboratory Manual". Failure to attend an allocated Laboratory session at the specified time will result in non-attendance being recorded and zero marks being allocated for the specific Laboratory component, unless documentary evidence of medical or other extenuating circumstances is

provided to the Laboratory Convenor (Dr Gretel Heber) within three (3) days of the laboratory session. Where satisfactory evidence is so provided, final attendance and laboratory marks will be proportionately adjusted to account for the approved absence(s).

**STUDENTS ARE REMINDED THAT PUNCTUALITY IS EXTREMELY IMPORTANT - THIS IS PARTICULARLY SO WITH LABORATORY EXPERIMENTS. IF STUDENTS ARE LATE, MARKS WILL BE DEDUCTED FROM THEIR MARK FOR THAT EXPERIMENT - THIS WILL BE EXPLAINED DURING LECTURES PRIOR TO STARTING THE LABORATORY SESSIONS**

### **End of Trimester Exam (45%)**

The purpose of this exam is to assess comprehension and application of the course material from Modules 1 - 4 by answer to multiple choice and problem based assessment items. Students are required to gain a mark of 40% or greater in this exam to pass the course.

### **Submission and Return of Assessment Items**

Examination papers will not be returned. Marked laboratory reports will be distributed in the laboratory classes.

### **Retention of Originals**

You must be able to produce a copy of all work submitted if so requested. Copies should be retained until after the release of final results for the course.

### **Extensions**

To apply for an extension of time for an assignment, you must submit an Application for Extension of Assignment form to your teacher at least 24 hours before the date the assignment is due. Grounds for extensions are usually: serious illness, accident, disability, bereavement or other compassionate circumstances and must be able to be substantiated with relevant documentation [e.g. Griffith College Medical Certificate]. Please refer to the Griffith College website - Policy Library - for guidelines regarding extensions and deferred assessment.

## Assessment Feedback

Laboratory performance will be graded on an ongoing basis.

Final grades for the course will be released as per normal Griffith College process.

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## Generic Skills

Griffith College aims to develop graduates who have an open and critical approach to learning and a capacity for lifelong learning. Through engagement in their studies, students are provided with opportunities to begin the development of these and other generic skills.

Studies in this course will give you opportunities to begin to develop the following skills:

Generic Skills	Taught	Practised	Assessed
Written Communication	Yes	Yes	Yes
Oral Communication	Yes	Yes	
Information Literacy		Yes	
Secondary Research			
Critical and Innovative Thinking			
Academic Integrity	Yes	Yes	Yes
Self-directed Learning			
Team Work	Yes	Yes	
Cultural Intelligence			
English Language Proficiency			

### *Additional Course Generic Skills*

Specific Skills	Taught	Practised	Assessed
Laboratory skills	Yes	Yes	Yes

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## Additional Course Information

Students should refer to the Griffith College portal for further information about this course.

Teacher and Course Evaluations

In response to students' feedback to make lectures more interactive, they are now encouraged to participate in lectures by both answering questions and trying examples of problems for themselves.

Your feedback is respected and valued by your lecturers and tutors. You are encouraged to provide your thoughts on the course and teaching, both positive and critical, directly to your lecturer and tutor or by completing course and lecturer evaluations on the Griffith College portal whenever these are available.

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## **Academic Integrity**

Griffith College is committed to maintaining high academic standards to protect the value of its qualifications. Academic integrity means acting with the values of honesty, trust, fairness, respect and responsibility in learning, teaching and research. It is important for students, teachers, researchers and all staff to act in an honest way, be responsible for their actions, and show fairness in every part of their work. Academic integrity is important for an individual's and the College's reputation.

All staff and students of the College are responsible for academic integrity. As a student, you are expected to conduct your studies honestly, ethically and in accordance with accepted standards of academic conduct. Any form of academic conduct that is contrary to these standards is considered a breach of academic integrity and is unacceptable.

Some students deliberately breach academic integrity standards with intent to deceive. This conscious, pre-meditated form of cheating is considered to be one of the most serious forms of fraudulent academic behaviour, for which the College has zero tolerance and for which penalties, including exclusion from the College, will be applied.

However, Griffith College also recognises many students breach academic integrity standards without intent to deceive. In these cases, students may be required to undertake additional educational activities to remediate their behaviour and may also be provided appropriate advice by academic staff.

As you undertake your studies at Griffith College, your lecturers, tutors and academic advisors will provide you with guidance to understand and maintain academic integrity; however, it is also your responsibility to seek out guidance if and when you are unsure about appropriate academic conduct.

In the case of any allegation of academic misconduct made against a student he or she may request the guidance and support of a Griffith College Student Learning Advisor or Student Counsellor.

Please ensure that you are familiar with the [Griffith College Academic Integrity Policy](#); this policy provides an overview of some of the behaviours that are considered breaches of academic integrity, as well as the penalties and processes involved when a breach is identified.

For further information please refer to the Academic Integrity Policy on the Griffith College website – Policy Library.

### **Risk Assessment Statement**

Laboratory safety training is compulsory prior to participate in laboratory classes.

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